

Name: _____

Homework from the book:

Ch 9 Multiple Choice: 1-10, 14-20, 24-27, 30-40 Exercises 34, 49-57(odd)

Study Guide : Multiple Choice: 8-11, Solved problems 9-1 thru 9-4

Questions on balancing reactions should be held until next week.

Take home quiz:

1) What is the total number of atoms in 1 $\text{Mg}(\text{ClO}_3)_2$? (A) 3. (B) 5. (C) 8. (D) 9. (E) 10

2) How many oxygen atoms are there in $\text{Pb}(\text{NO}_3)_2$? (A) 1. (B) 2. (C) 3. (D) 5. (E) 6.

3) The formula of carbon tetrabromide is (A) CBr . (B) CBr_4 . (C) C_4Br . (D) CBr_4 . (E) C_2Br_4 .

4) What is the chemical formula for calcium chloride? _____

5) What is the chemical formula for Iron(II) hydroxide? _____

6) What is the name of the compound with the chemical formula Li_2CO_3 ? _____

7) What is the chemical formula for copper (II) bromide? _____

8) What is the name of the compound with the chemical formula CO_2 ? _____

9) What is the chemical formula for ammonium acetate? _____

10) What is the chemical formula for sodium sulfate? _____