## CHM1032C

## **Test 2 Homework**

- Know the names and symbols for the first 36 elements
- Memorize the 10 polyatomic ions discussed in class.
- Be able to draw the Lewis structures for the 10 polyatomic ions you were asked to memorize.

Chapter 3: 27, 29, 31, 33, 37, 39 abd, 41, 43, 45, 51, 53, 59, 61, 65-67, 69-77, 79-85, 93, 95, 97

Extra Problems:

- 1. What is the name of the compound with the chemical formula NH<sub>4</sub>NO<sub>3</sub>?
- 2. What is the name of the compound with the chemical formula CrO<sub>3</sub>?
- 3. What is the chemical formula for copper(II) cyanide?

Chapter 4: 27, 31, 33, 35-42, 53-61, 67, 69, 70, 73, 75, 77, 79, 89

Extra Problems:

- 1. What is the chemical formula for phosphorous trichloride?
- 2. Lewis Structures: Please draw the Lewis structure for the polyatomic ion NO<sub>3</sub>. *Please show all formal charges.*

What is the electron pair geometry about the central N?\_\_\_\_\_\_
What are the bond angles about the central N?\_\_\_\_\_\_

Chapter 11: 23, 25, 27, 29, 31, 35-39, 41, 45-49, 53, 58, 59, 83

**Chapter 12**:33, 35, 37, 38, 43, 45, 47, 49, 51, 53, 57, 59

Extra Problem: Draw and name all of the isomers for C<sub>7</sub>H<sub>16</sub>.